Iodine Value of a Fixed Oil

Experiment 18 - Practical Notes

Name: _____ Roll no.: _____ Date: _____

Aim

To determine iodine value of the assigned oil by the specified iodine-value method.

Principle

The assigned halogen-addition method measures unsaturation through the difference between blank and test titration.

Requirements

- Assigned test material and the prescribed method reagents or standards.
- Calibrated volumetric glassware, appropriate instrument and record sheet.

Procedure

1. Prepare and identify the test

Confirm the analyte, method version, concentration basis and test-portion amount. Prepare the test solution as specified and record any dilution.

2. Set the measurement basis

Check the instrument or standardised titrant. Prepare the specified blank, standard and control so that the test and comparison use the same basis.

3. Make the measurement

Apply the prescribed the specified iodine-value method method and record blank and test thiosulphate titres. Record replicate readings and any interfering appearance.

4. Calculate

Apply the stated equation with the correct blank correction, dilution and unit conversion. Show the substitution and retain the observed readings.

5. Assess the result

Compare replicate agreement and the assigned specification. Record a result with units and the limitations of the method.

Observations and Record

Measurement	Reading
Test portion / dilution	_____
Blank / standard	_____
blank and test thiosulphate titres	_____
Calculated content and unit	_____

Worked Example

The numbers below are illustrative; enter your own observations in the record.

- Iodine value = (B − S) × N × 12.69 / m; B and S in mL, N in equivalents/L, m in g. The unit is g iodine equivalent per 100 g oil.
- Example: B − S = 10.0 mL, N = 0.1 and m = 0.2 g gives 63.45.

Result

Record the completed observations, calculations and conclusion with units. State any limitation of the observation.

Precautions

- Use the exact method reagents and assigned endpoint; different methods are not interchangeable.
- Record the blank correction and the actual test portion rather than an assumed value.
- Hazardous reagent operations require the specified laboratory facilities and supervision.

Viva Questions and Answers

1. What produces the analytical signal?

The assigned halogen-addition method measures unsaturation through the difference between blank and test titration.

2. Why include a blank or control?

It identifies contributions unrelated to the analyte or checks that the measurement system performed appropriately.

3. What must accompany the result?

Units, method, dilution basis, relevant control performance and any limitation.

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